



## The College of Graduate Studies and the College of Engineering Cordially Invite You to a

## **Master Thesis Defense**

<u>Title</u> DESALINATION WASTE REVITALIZATION USING CARBON CAPTURE: PROCESS OPTIMIZATION USING MACHINE LEARNING

Вy

Ahmed Mohamed Nasereldin Mohamed Elsayed <u>Faculty Advisor</u> Prof. Ali H. Al-Marzouqi and Prof. Nazar Zaki <u>Date and Time:</u> 11:00 AM Thursday, 19 May 2022 F3-40 Abstract

Over the next five to ten years, desalination will play an ever-important role in our society. Developing nations will look at desalination from the point of water scarcity, while developed nations will consider the perspective of the Water-Energy-Food nexus. With current technology, every liter of freshwater produced, one and a half liter is thrown away as waste. This body of research proposes to employ chemical precipitation in ammoniated brine solutions, that have reasonably high ionic concentrations.  $CO_2$  absorption in Bubble Column Reactors can reduce sodium ions by precipitating NaHCO<sub>3</sub>. Also, the absorbed  $CO_2$  can reduce chloride ions and induce hydrate formation, by operating the column at a lower temperature (10 - 15 °C).

The Bubble Column Reactor model includes  $CO_2$  absorption, aqueous reaction, and salt precipitation. The kinetic/equilibrium reaction system is simulated and optimized using nonlinear programming. The hydrate formation process is modeled and optimized using Machine Learning. The results from the optimization show that brine with 23400 ppm of Na<sup>+</sup> ions and 44000 ppm of Cl<sup>-</sup> ions can be reduced by 72.5% and 54.2%, respectively. The absorption process will use 1.15 mol of CO<sub>2</sub> absorbed per mole of NH<sub>3</sub> dissolved with a total of 8.702 mol of CO<sub>2</sub> absorbed.